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NEXT PAGE. -7- CH 3CH 2COOH(aq) + H 2O(1) CH 3CH 2COO (aq) + H 3O+(aq) 2. Propanoic acid, CH 3CH 2COOH, is a carboxylic acid that reacts with water according to the equation above. At

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(d) Calculate the number of moles of precipitate that is produced in the experiment. 1.698 g 1.462 g = 0.236 g PbI 2 (s) 2 4 22 2 1 mol PbI 0.236 g PbI =  $5.12\ 10\ mol\ PbI$ 

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