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Description. Now in its fifth edition, Housecroft & Sharpe's Inorganic Chemistry is a well-respected and leading international textbook. This Solutions Manual accompanies the main text and provides model answers to the end-of-chapter problems, linking to relevant sections and figures in the main text as appropriate.

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Note : # Neutrons = Mass Number – Atomic Number. # Neutrons of Nitrogen-14 = 14 - 7 = 7. # Neutrons of Nitrogen-13 = 13 - 7 = 6. c) The number of electrons in a neutral atom always equals the atomic number. The atomic number of both isotopes is 7, therefore the number of electrons in both isotopes is also 7.

Organic Chemistry, 5th Edition Chapter 1 - Section 1.1 ...

(Northern Arizona University) and Raymond Chang, this success guide is written for use with General Chemistry. It aims to help students hone their analytical and problem-solving skills by presenting detailed approaches to solving chemical problems.

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This latest edition of the most internationally respected reference in food chemistry for more than 30 years, Fennema's Food Chemistry, 5th Edition once again meets and surpasses the standards of quality and comprehensive information set by its predecessors. All chapters reflect recent scientific advances and, where appropriate, have expanded and evolved their focus to provide readers with the current state-of-the-science of chemistry for the food industry. This edition introduces new editors and contributors who are recognized experts in their fields. The fifth edition presents a completely rewritten chapter on Water and Ice, written in an easy-to-understand manner suitable for professionals as well as undergraduates. In addition, ten former chapters have been completely revised and updated, two of which receive extensive attention in the new edition including Carbohydrates (Chapter 3), which has been expanded to include a section on Maillard reaction; and Dispersed Systems: Basic considerations (Chapter 7), which includes thermodynamic incompatibility/phase separation concepts. Retaining the straightforward organization and accessibility of the original, this edition begins with an examination of major food components such as water, carbohydrates, lipids, proteins, and enzymes. The second section looks at minor food components including vitamins and minerals, colorants, flavors, and additives. The final section considers food systems by reviewing basic considerations as well as specific information on the characteristics of milk, the postmortem physiology of edible muscle, and postharvest physiology of plant tissues.

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Some printings include access code card, "Mastering Chemistry."

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